

Enehydroxylamines as Versatile Compounds in 3,3-Sigmatropic Rearrangements†

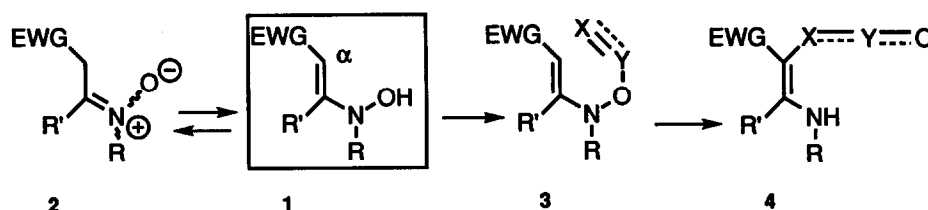
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Abstract: Enehydroxylamines react readily, in the presence of base, with electrophiles containing unsaturation to afford intermediates that, either spontaneously or upon heating, lead via a 3,3-sigmatropic rearrangement to α -substituted products.

Enehydroxylamines **1**, tautomers of the nitrones **2**, are compounds with an interesting chemistry, owing to the presence in the same molecule of an enaminic nitrogen and a weak N-O bond. These compounds have been frequently postulated as intermediates in reactions of certain nitrones with acyl chlorides but seldom isolated.^{1,2}



EWG = electron - withdrawing group

Scheme 1

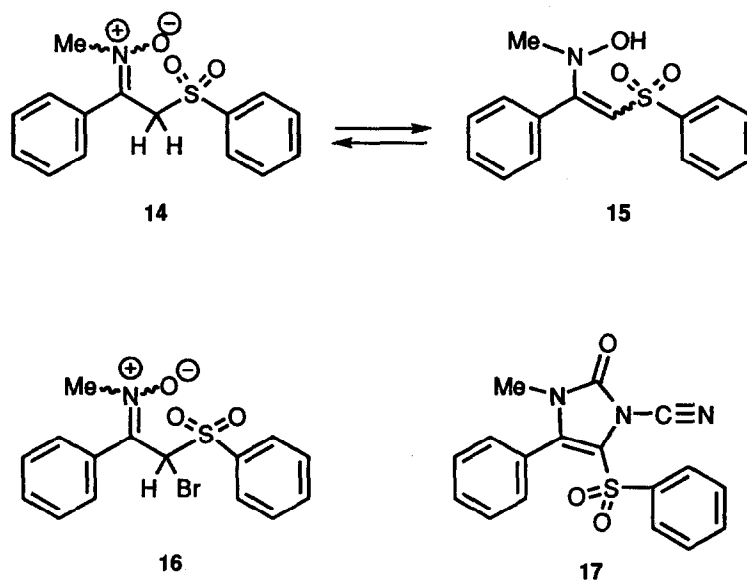
We report in this communication³ results of our study of representatives of **1** (Scheme 1) with a variety of electrophiles, where facile α -substitution, as depicted in **4**, was observed, most probably through a 3,3-sigmatropic rearrangement of the initially formed unstable intermediate **3**. Compounds **5a** and **5b** (Scheme 2) could be easily prepared by reaction of N-methyl hydroxylamine with the appropriate cyclic 1,3-diketone in toluene at room temperature.^{4,5} Reaction of **5a** with cyanogen bromide (1 eq. or 3 eq.) in the presence of DABCO, in THF, at room temperature (4 h) yielded only **6a** in 68% yield (Scheme 2, equation 1). Similarly **5b** led to **6b** (83%). However with the more powerful electrophile, 1-cyano-4-N,N-dimethylaminopyridinium

† This paper is dedicated with admiration to Professor Sir Derek Barton on the occasion of his 75th birthday.

bromide **6** (CAP) (3 eq.), and *N,N*-diisopropyl-*N*-ethylamine as the base, the *ciano* compounds **7a** or **7b** were obtained (equation 2). Other electrophiles such as PhCOCl (equation 3), MeSO₂Cl (equation 4) and *N,N*-dimethylthiocarbamoyl chloride (equation 5), reacted smoothly with **5a** or **5b**, in the presence of base, and afforded products **8** (a,b), **9** (a,b) and **10** (a, b) respectively in good to excellent yields. The carbinolamines **11a** and **11b**, obtained from **5a** and **5b**, respectively, with methyl propiolate, were smoothly dehydrated (toluene, Δ) to yield the synthetically useful pyrrole derivatives **12a**⁷ and **12b** (equation 6). With phenyl isocyanate (2.5 eq.) **5a** yielded **13a** (equation 7) through a sequence of reactions involving the formation of a carbamate derivative, followed by rearrangement, decarboxylation of the resulting *N*-phenyl carbamic acid and subsequent reaction of the product formed with a second equivalent of the electrophile.

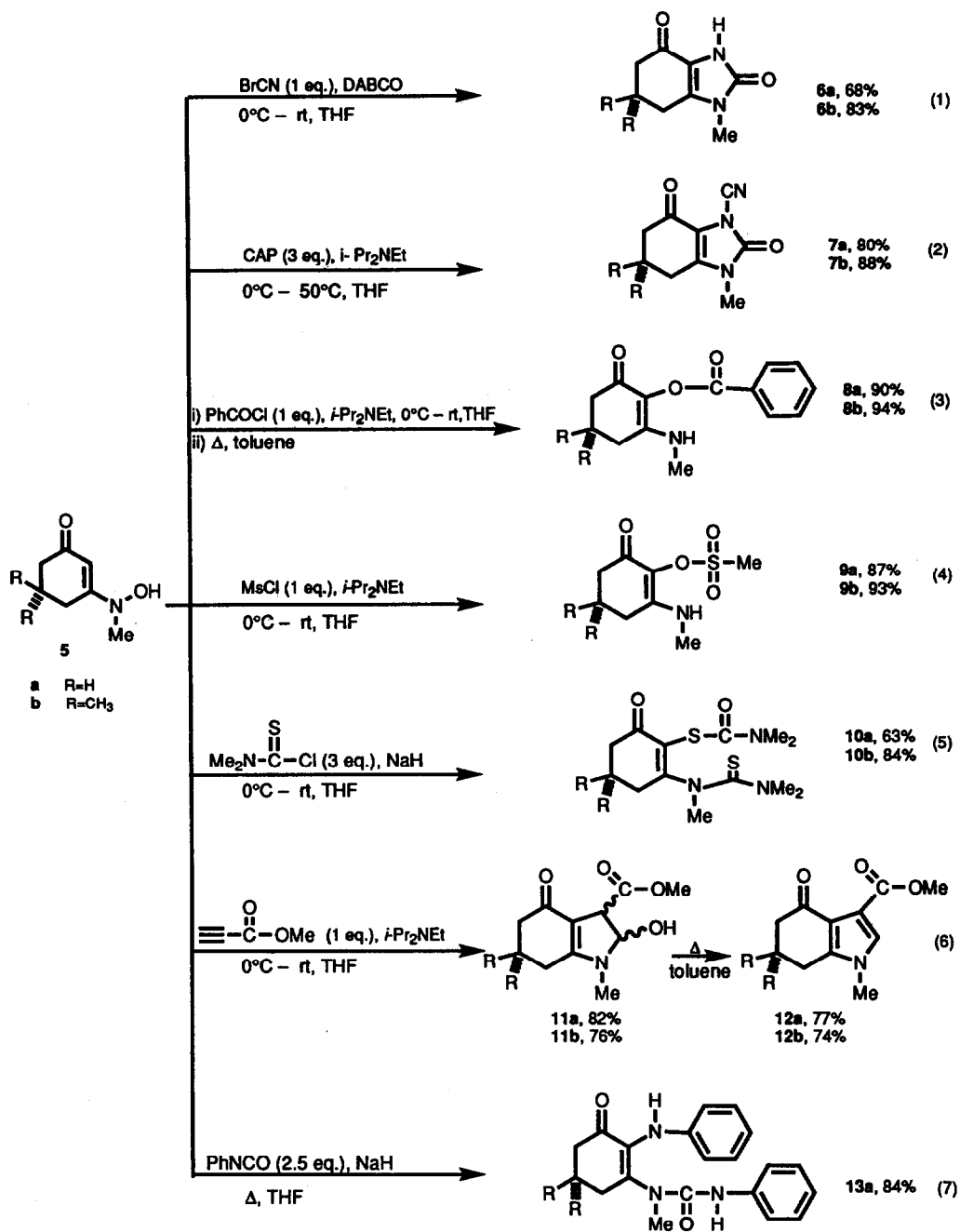
Simple open chain nitrones, such as the one derived from acetophenone and *N*-methyl hydroxylamine failed to give any useful products. However, the nitron **14**, while yielding the bromo compound **16** (69%) with BrCN⁸ and base, generated the expected imidazolone derivative **17** in 80% yield *via* **15**, when treated successively with Hunig's base and CAP.

In conclusion, it is shown that cyclic enehydroxylamines are useful in 3,3-sigmatropic reactions, whereby a sulphur, an oxygen, a nitrogen and a carbon atom can be selectively introduced at the α-position by the appropriate choice of reagents, under mild conditions, to provide cyclohexane molecules with three contiguous but different functional groups.



ACKNOWLEDGMENTS

We thank Junta Nacional de Investigação Científica e Tecnológica (JNICT, Lisbon) for partial financial support and for the award of a post-graduate fellowship to one of us (L. V. R.).



Scheme 2

REFERENCES AND NOTES

1. Cummins, C. H.; Coates, R. M. *J. Org. Chem.*, **1983**, *48*, 2070-2076, and references therein; Coates, R. M.; Cummins, C. H. *J. Org. Chem.*, **1986**, *51*, 1383-1389.
2. Blechert, S. *Synthesis*, **1989**, 71-82.
3. This work was presented in part at the VIIth FECHEM Conference on Heterocycles in Bio-organic Chemistry, Santiago de Compostela (Spain), 26-29th September, 1993.
4. A mixture of N-methyl hydroxylamine HCl (0.036 eq.) and Et₃N (0.036 eq.) in Na-dried toluene, after being stirred vigorously under N₂ at room temperature, was filtered and the filtrate treated dropwise with a toluene solution of 1,3-cyclohexanedione (0.024 eq.). After 1,5 h the solvent was removed *in vacuo* and the residue crystallised to yield pure **5a** (59%); m.p. 126-128 °C (EtOAc-MeOH); selected NMR data δ (CDCl₃) 3.350 (3H, s, Me), 5.460 (1H, s, H-2, D₂O exchange), 7.617 (1H, br s, D₂O exchange); m/z (%) 141 [M⁺] (85), 125 [M⁺-16] (26), 113 (100). **5b** : m.p. 158-160 °C (CHCl₃-n-hexane) (lit.⁵ m.p. 158-160 °C).
6a : m.p. 241-242 °C (dec.) (Et₂O-CH₂Cl₂); **6b** : m.p. 236-238 °C (dec.) (Et₂O-CH₂Cl₂); **7a** : m.p. 204-205 °C (EtOAc); **7b** : m.p. 184-185 °C (CH₂Cl₂-EtOAc); **8a** and **8b** oils; **9a** : m.p. 147-148 °C (Et₂O-CH₂Cl₂); **9b** : m.p. 141-142 °C (Et₂O-CH₂Cl₂); **10a** : m.p. 133-134 °C (Et₂O-CH₂Cl₂); **10b** : m.p. 125-126 °C (Et₂O-CH₂Cl₂); **11a** : oil; **11b** : m.p. 147-149 °C (dec.) (Et₂O-CH₂Cl₂); **12a** : m.p. 167-168 °C (Et₂O-CH₂Cl₂); **12b** : m.p. 142-143 °C (Et₂O-CH₂Cl₂); **13a** : m.p. 225-228 °C (dec.) (EtOAc-MeOH); **14** : m.p. 139-141 °C (Et₂O-CH₂Cl₂); **16** : m.p. 130-131 °C (Et₂O-CH₂Cl₂); **17** : m.p. 177-178 °C (EtOH).
 All new compounds showed satisfactory spectroscopic data together with microanalysis and/or mass spectrometry data.
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6. Whitten, J. P.; McCarthy, J. R.; Matthews, D. P. *Synthesis*, **1988**, 470-472.
7. Compounds of this type have been extensively used for the synthesis of 4-substituted indole derivatives: Matsumoto, M.; Watanabe, N. *Heterocycles*, **1984**, *22*, 2313-2316; Hatanaka, N.; Matsumoto, M. *Heterocycles*, **1986**, *24*, 1963-1971; Hatanaka, N.; Watanabe, N.; Matsumoto, M. *Heterocycles*, **1986**, *24*, 1987-1996, and references therein.
8. Durst, T. In *Comprehensive Organic Chemistry*; Barton, D.; Ollis, W. D. Eds.; Pergamon: Oxford, 1979; vol. 3, p. 190.

(Received in UK 12 January 1994; revised 10 February 1994; accepted 18 February 1994)